# Synthesis of Heterocyclic Compounds. Part 46.' The Reactions of Malonamide and 2-Cyanoacetamide with Substituted Propenones 

José L. Soto,* Carlos Seoane, Pilar Zamorano, Maria J. Rubio, Alejandro Monforte, and Margarita Quinteiro<br>Departamento de Química Orgánica, Facultad de Química, Universidad Comp/utense, 28040-Madrid, Spain


#### Abstract

The synthesis of substituted 2-pyridone derivatives from malonamide (1) and 2-cyanoacetamide (2) is reported. Conjugate addition of these compounds to substituted propenones (3) and (4) leads to 6 -hydroxy-2-piperidones, which can be dehydrated and aromatized to 2 -pyridones (7), (11), and (14). The intermediate 3,4-dihydro derivatives were also obtained and some chemical transformations of the products prepared are also described.


In a previous communication, ${ }^{2}$ we described a modification of the Guareschi pyridone synthesis, ${ }^{3.4}$ involving the conjugate addition of 2 -cyanoacetamide to $\alpha$-benzoylcinnamonitriles, and leading to some cyanopyridones which were also obtained by ring transformations of aminopyran derivatives. ${ }^{5}$ We now report the extension of this synthesis to different $\alpha, \beta$ unsaturated ketones and to the use of malonamide, as well as some chemical transformations of the resulting 2 -pyridones.
In contrast to the Guareschi synthesis, conjugate addition to an unsaturated ketone leads to a saturated intermediate (see Scheme), which does not provide an aromatic ring upon dehydration. Thus, addition of malonamide (1) to $\alpha$-benzoylcinnamonitriles (3), which takes place in a similar way to the reaction with 2 -cyanoacetamide, is followed only by a cyclization to a piperidine ring. The reaction is very easy to perform in basic medium and affords 6 -hydroxy-2-piperidones (5) in good yields with a variety of substituents on the aryl groups. Dehydration of hydroxy derivatives (5) must be carried out in a separate step by using dilute sulphuric acid and leads to 3,4-dihydro-2-pyridones (6), which result also from treatment of compounds(9) with concentrated sulphuricacid. Transformation of compounds (6) into dicarbonitriles (10) can be achieved by reaction with phosphoryl trichloride in dimethylformamide (DMF), and the reverse transformation takes place upon acid treatment. It must be pointed out that only the cyano group located at position 3 of compounds (10) is hydrolysed, whereas the nitrile at position 5 does not react. This is in agreement with the known facility of hydrolysis of a cyano group contiguous to the carbonyl group in a pyridone ring. ${ }^{6.7}$
Aromatic 2 -pyridones (7) result from oxidation of their 3,4dihydro derivatives (6) with nitrosylsulphuric acid. Alternatively, this reagent can be employed to bring about the direct transformation of 6 -hydroxy-2-piperidones (5) into 2 -pyridones (7). In both cases, the reaction is very easy to perform and the yields obtained are high. On the other hand, the carbamoyl group of 2-pyridones (7) can be removed through hydrolysis to a carboxyl group and decarboxylation. In fact, treatment of compounds of (7) with concentrated hydrochloric acid in dioxane brings about their transformation into 2 -pyridones (8), unsubstituted at position 3. Isolation of the intermediate 2-oxopyridine-3-carboxylic acid was not possible. Compounds (8) were also obtained from 3,5-dicyano-2-pyridones (11) by treatment with hydrochloric acid in ethylene glycol. In both reactions, despite the long reaction time (several days), the cyano group located at position 5 is unaffected.
The reactions of malonamide (1) and 2-cyanoacetamide (2) with $\alpha$-benzoylcinnamates (4) were of interest because, in this case, cyclization could take place by attack of the amide nitrogen either at the carbonyl group or at the ethoxy carbonyl group. We found that, with 2 -cyanoacetamide, the cyclization involves attack at the carbonyl group, leading to 2 -piperidones
(12), whereas no reaction took place between malonamide and ethyl $\alpha$-benzoylcinnamates (4) under a variety of conditions. This is probably due to the lower acidity of the methylene group of malonamide and the lower susceptibility of ethyl $x$ benzoylcinnamates to Michael attack in comparison with $\alpha$ benzoylcinnamonitriles. Preparation of piperidones (12) requires a longer reaction time than that for piperidones (9), but they are obtained in good yield. They are rather unstable compounds and decompose easily on heating. Furthermore, the reaction fails when a methoxy group is present on the starting material [ethyl $\alpha$-benzoylcinnamate (4d)], because the electrondonating effect of the methoxy group makes conjugate addition more difficult.
Compounds (12) can be dehydrated to 3,4-dihydropyridone derivative (13) by treatment with $75 \%$ sulphuric acid at room temperature. The reaction gives good yields, but the acid concentration is critical. If $50 \%$ sulphuric acid is used no reaction occurs, and with $98 \%$ sulphuric acid dehydration is accompanied by hydrolysis of the ethoxycarbonyl group thus leading to carboxylic acids ( 15 ), which can also be obtained by similar treatment applied to the 3,4-dihydro-2-pyridones (13). On the other hand, 6 -hydroxy-2-piperidones (12) can be transformed into 2 -pyridones (14) by treatment with thionyl chloride in pyridine at $-10^{\circ} \mathrm{C}$. According to the known behaviour of this reagent with other heterocyclic systems, ${ }^{8.9}$ it effects dehydration and dehydrogenation, in one single step, of the aromatic system (14), which can also be prepared from 3,4dihydro derivatives (13) by reaction with nitrosylsulphuric acid.
Spectral data of the new compounds prepared (see Experimental section) are in agreement with reported values for other 2 -pyridones. ${ }^{10-12}$
Finally, in the n.m.r. spectra of the 3,4-dihydro-2-pyridones obtained, the hydrogens located at positions 3 and 4 of the ring appear as two doublets at $\delta_{\mathrm{H}} 4.2$ and 3.6 in compounds (6) and at $\delta_{H} 5.0$ and 4.4 in compounds (13). The coupling constant measured for compounds (6) is 7 Hz and whereas that for compounds (13) is 6 Hz . When the Haasnoot-Leeuw-Altona equation ${ }^{13}$ is applied to both compounds, taking into account the correction for the Huggins ${ }^{14}$ electronegativities of the groups attached to carbons 3 and 4 of the ring, a dihedral angle of about $40^{\circ}$ is obtained. This result indicates an axialequatorial geometry for the coupled protons.

## Experimental

M.p.s were determined with a Büchi apparatus for samples in capillary tubes and are uncorrected. I.r. spectra were recorded with a Perkin-Elmer 257 spectrophotometer for samples in potassium bromide pellets and ${ }^{1} \mathrm{H}$ n.m.r. spectra were measured at 60 MHz on a Varian T-60A spectrometer for $\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}$ solutions; chemical shifts are given in $\delta$ values against $\mathrm{SiMe}_{4}$ as

(15)

Scheme. Reagents: i, piperidine-dry $\mathrm{MeOH} ; \mathrm{ii}, 50 \% \mathrm{H}_{2} \mathrm{SO}_{4} ;$ iii, $98 \% \mathrm{H}_{2} \mathrm{SO}_{4}$; iv, $\mathrm{HOSO}_{2} \mathrm{ONO}$-aq. AcOH; v, $\mathrm{POCl}_{3}$-DMF; vi, aq. HCl -dioxane; vii, aq. $\mathrm{HCl}-\mathrm{HOCH}_{2} \mathrm{CH}_{2} \mathrm{OH}$; viii, $75 \% \mathrm{H}_{2} \mathrm{SO}_{4}$; ix, $\mathrm{Cl}_{2} \mathrm{SO}-$ pyridine
internal standard. A Varian MAT 711 spectrometer was used for recording mass spectra. Microanalyses were performed by Centro Nacional de Química Orgánica de Madrid. The reactions and purity of compounds were monitored by t.l.c., performed on silica gel plates (Merck) and using toluene-ethyl acetate as developer.

Malonamide, 2-cyanoacetamide, and ethyl benzoylacetate were commercially obtained and were used without further purification. $\alpha$-Benzoylcinnamonitriles (3) ${ }^{15.16}$ and ethyl $\alpha$ benzoylcinnamates (4) ${ }^{17-20}$ were prepared according to known methods as was $\omega$-cyanoacetophenone. ${ }^{21}$

4,6-Diaryl-3-carbamoyl-5-cyano-6-hydroxypiperidin-2-ones
(5). General Procedure.-To a suspension of the appropriate $\alpha$ benzoylcinnamonitrile (3) ( 5 mmol ) in dry methanol ( 5 ml ) was added a solution of malonamide (1) ( 6 mmol ) in dry methanol ( 12 ml ), together with a few drops of piperidine. The reaction mixture was stirred at room temperature for $14-16 \mathrm{~h}$. After that time, no starting material remained (t.l.c.) and the solid that precipitated was collected by filtration, washed with cold methanol and purified by recrystallization in methanol.

3-Carbamoyl-5-cyano-6-hydroxy-4,6-diphenylpiperidin-2-one (5a). This compound was obtained in $86 \%$ yield. M.p. 206-207 ${ }^{\circ} \mathrm{C}$ (Found: C, 68.2; H, 5.1; N, 12.2. $\mathrm{C}_{19} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{3}$ requires $\mathrm{C}, 68.04 ; \mathrm{H}, 5.10 ; \mathrm{N}, 12.53 \%$ ); $v_{\text {max. }} 3480,3360,3210$, $3100,2910,2230,1660,1590$, and $1480 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 8.30(1 \mathrm{H}, \mathrm{s}$,
$\mathrm{NH}), 6.50-7.53\left(13 \mathrm{H}, \mathrm{m}, \mathrm{ArH}, \mathrm{OH}\right.$, and $\left.\mathrm{NH}_{2}\right)$, and $3.43-4.03$ ( $3 \mathrm{H}, \mathrm{m}$ ).

3-Carbamoyl-5-cyano-6-hydroxy-6-phenyl-4-(p-tolyl)-piperidin-2-one (5b). $82 \%$ yield, m.p. $159-160^{\circ} \mathrm{C}$ (Found: C, 68.9; $\mathrm{H}, 5.55 ; \mathrm{N}, 11.8 . \mathrm{C}_{20} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{3}$ requires $\mathrm{C}, 68.75 ; \mathrm{H}, 5.48$; N , $12.02 \%$ ); $v_{\max } 3400,3280,3220,3120,2250,1670,1620$, 1515 , and $1455 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 8.36(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 6.50-7.60(12 \mathrm{H}, \mathrm{m}$, $\mathrm{ArH}, \mathrm{OH}$, and $\mathrm{NH}_{2}$ ), 3.30-4.10 ( $3 \mathrm{H}, \mathrm{m}$ ), and $2.20(3 \mathrm{H}, \mathrm{s}$, $\mathrm{CH}_{3}$ ).
3-Carbamoyl-4-(p-chlorophenyl)-5-cyano-6-hydroxy-6-phenylpiperidin-2-one (5c). $80 \%$ yield, m.p. $169-170^{\circ} \mathrm{C}$ (Found: C, 61.4; H, 4.5; N, 11.3; Cl, 10.1. $\mathrm{C}_{19} \mathrm{H}_{16} \mathrm{ClN}_{2} \mathrm{O}_{3}$ requires C, $61.70 ; \mathrm{H}, 4.36 ; \mathrm{N}, 11.36 ; \mathrm{Cl}, 9.58 \%$ ), $\mathrm{v}_{\text {max. }} 3420,3360,3260$, $3200,3090,2260,1675,1660,1610,1490$, and $1440 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}}$ $8.43(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 6.60-7.60\left(12 \mathrm{H}, \mathrm{m}, \mathrm{ArH}, \mathrm{OH}\right.$, and $\left.\mathrm{NH}_{2}\right)$, and $3.30-3.40(3 \mathrm{H}, \mathrm{m})$.
3-Carbamoyl-5-cyano-6-hydroxy-4-(p-methoxyphenyl)-6-
phenylpiperidin-2-one (5d). $87 \%$ yield, m.p. $155-156{ }^{\circ} \mathrm{C}$ (Found: C, 65.3; H, 5.6; N, 11.75. $\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{4}$ requires C, 65.74; H, 5.24; N, 11.50\%); $v_{\text {max. }} 3400,3300,3210,3130,2240$, $1670,1620,1580,1510$, and $1460 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 8.50(1 \mathrm{H}, \mathrm{s}, \mathrm{NH})$, $6.70-7.8012 \mathrm{H}, \mathrm{m}, \mathrm{ArH}, \mathrm{OH}$, and $\left.\mathrm{NH}_{2}\right), 3.80-4.60(3 \mathrm{H}, \mathrm{m})$, and $3.70\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{O}\right)$.

3-Carbamoyl-5-cyano-6-hydroxy-4-(p-nitrophenyl)-6-phenylpiperidin-2-one (5e). $63 \%$ yield, m.p. $155-156{ }^{\circ} \mathrm{C}$ (Found: C, 59.6; $\mathrm{H}, 4.1 ; \mathrm{N}, 14.7 . \mathrm{C}_{19} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}_{5}$ requires C, 59.99; $\mathrm{H}, 4.24$;
$\mathrm{N}, 14.63 \%$ ); $v_{\text {max. }} 3400,3260,3200,3100,2930,2240,1665$, $1615,1590,1515$, and $1490 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 8.70(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 6.70-$ $8.30\left(12 \mathrm{H}, \mathrm{m}, \mathrm{ArH}, \mathrm{OH}\right.$, and $\left.\mathrm{NH}_{2}\right)$, and $3.50-4.50(3 \mathrm{H}, \mathrm{m})$.

4,6-Diaryl-3-carbamoyl-5-cyano-3,4-dihydropyridin-2(1H)ones (6). General Procedure.-A suspension of the corresponding 6-hydroxypiperidin-2-one (5) ( 2 mmol ) in $50 \%$ sulphuric acid $(20 \mathrm{ml})$ was stirred at room temperature for $3-6 \mathrm{~h}$. The reaction mixture was then poured into crushed ice and the resulting solid was collected by filtration and washed with plenty of water, to remove any remaining acid, until neutral pH was obtained. Compounds (6) thus obtained were recrystallized in methanol.

3-Carbamoyl-5-cyano-4,6-diphenyl-3,4-dihydropyridin-2-
(1H)-one ( 6 a ). $84 \%$ yield, m.p. $225-226^{\circ} \mathrm{C}$ (Found: C, 72.3 ; H, 4.7; $\mathrm{N}, 13.1 . \mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2}$ requires $\mathrm{C}, 71.90 ; \mathrm{H}, 4.76 ; \mathrm{N}, 13.24 \%$ ), $v_{\text {max. }} 3450,3340,3220,3130,2220,1700,1680,1635,1600$, 1575 , and $1490 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 10.43(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 6.90-7.50(12 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{NH}_{2}\right), 4.13-4.33(1 \mathrm{H}, \mathrm{d}$, and $3.50-3.70(1 \mathrm{H}, \mathrm{d})$.

This compound was also obtained by using concentrated sulphuric acid, but the yield was only $75 \%$.

3-Carbamoyl-5-cyano-6-phenyl-4-(p-tolyl)-3,4-dihydropyri-$\operatorname{din}-2(1 \mathrm{H})$-one ( 6 b ). $65 \%$ yield, m.p. $180-181^{\circ} \mathrm{C}$ (Found: C , $72.0 ; \mathrm{H}, 5.6 ; \mathrm{N}, 12.4 . \mathrm{C}_{20} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{2}$ requires $\mathrm{C}, 72.48 ; \mathrm{H}, 5.17$; N , $12.68 \%$ ); $v_{\text {max. }} 3660,3480,3400,3320,2220,2200,1700$, $1670,1620,1510$, and $1480 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 10.43(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 6.90-$ $7.60\left(11 \mathrm{H}, \mathrm{m}, \mathrm{ArH}\right.$ and $\left.\mathrm{NH}_{2}\right), 4.10-4.30(1 \mathrm{H}, \mathrm{d}), 3.50-3.70(1$ $\mathrm{H}, \mathrm{d}$ ), and $2.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right)$.

3-Carbamoyl-4-(p-chlorophenyl)-5-cyano-6-phenyl-3,4-dihydropyridin- $2\left(1 \mathrm{H}\right.$ )-one ( $\mathbf{6 c}$ ). $84 \%$ yield, m.p. $173-174{ }^{\circ} \mathrm{C}$ (Found: C, 64.6; H, 4.2; N, 11.7; Cl, 10.0. $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{ClN}_{3} \mathrm{O}_{2}$ requires $\mathrm{C}, 64.86 ; \mathrm{H}, 3.98 ; \mathrm{N}, 11.94 ; \mathrm{Cl}, 10.09 \%$ ); 3630,3390 , $3320,3180,2220,2$ 200, $1695,1665,1615,1590$, and 1490 $\mathrm{cm}^{-1} ; \delta_{\mathrm{H}} 10.46(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 6.80-7.60(11 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$, and $\left.\mathrm{NH}_{2}\right), 4.13-4.36(1 \mathrm{H}, \mathrm{d})$, and $3.50-3.74(1 \mathrm{H}, \mathrm{d})$.

3-Carbamoyl-5-cyano-4-(p-methoxyphenyl)-6-phenyl-3,4-dihydropyridin- $2\left(1 \mathrm{H}\right.$ )-one ( $\mathbf{6 d}$ ). $68 \%$ yield, m.p. $197-198{ }^{\circ} \mathrm{C}$ (Found: C, 69.0; H, 5.0; N, 12.3. $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{3}$ requires C , 69.15; H, 4.93; N, 12.09\%); $v_{\text {max }} 3640,3400,3$ 320, 3 200, 3140,2220 , $2200,1685,1665,1665,1610,1590,1510,1490,1470$, and $1440 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 8.60(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 6.40-7.20(9 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$, and $\left.\mathrm{NH}_{2}\right), 4.00-4.20\left(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{3}\right), 3.40-3.70(1 \mathrm{H}, \mathrm{d})$, and $3.40(3$ $\mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{O}$ ).

3-Carbamoyl-5-cyano-4-(p-nitrophenyl)-6-phenyl-3,4-dihydropyridin- $2\left(1 \mathrm{H}\right.$ )-one ( 6 e ). $62 \%$ yield, m.p. $274-275{ }^{\circ} \mathrm{C}$ (Found: C, 63.1; H, 4.1; $\mathrm{N}, 15.5 . \mathrm{C}_{19} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{4}$ requires $\mathrm{C}, 62.98$; $\mathrm{H}, 3.86 ; \mathrm{N}, 15.46 \%$ ); $v_{\text {max. }} 3410,3300,3200,3100,2200,1690$, $1665,1600,1570,1505$, and $1490 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 10.53(1 \mathrm{H}, \mathrm{s}, \mathrm{NH})$, $7.00-8.20\left(11 \mathrm{H}, \mathrm{m}, \mathrm{ArH}\right.$, and $\left.\mathrm{NH}_{2}\right), 4.40-4.60(1 \mathrm{H}, \mathrm{d})$, and $3.63-3.83(1 \mathrm{H}, \mathrm{d})$.

Transformation of 3,4-Dicyano-6-hydroxy-4,6-diphenyl-piperidin-2-one (9a) into 3-Carbamoyl-5-cyano-4,6-diphenyl-3,4-dihydropyridin-2(1H)-one (6a). -The piperidinone (9a) (0.3 mmol ) was dissolved in concentrated sulphuric acid ( 1 ml ) and the solution was kept for 24 h . The mixture was then poured into crushed ice and the precipitate was collected by filtration and washed with water. Compound (6a) thus obtained in $24 \%$ yield was recrystallized from methanol.

Transformation of 3,5-Dicyano-4,6-diphenyl-3,4-dihydropy-ridin- $2(1 \mathrm{H}$ )-one (10a) into 3-Carbamoyl-5-cyano-4,6-diphenyl-3,4-dihydropyridin- $2(1 \mathrm{H}$ )-one ( $\mathbf{6 a}$ ).-A solution of compound (10a) ( 0.7 mmol ) in concentrated sulphuric acid ( 2 ml ) was kept at room temperature for 20 h and was then poured into crushed ice. The collected precipitate is washed with water and recrystallized from methanol. Yield of (6a) $40 \%$.

Transformation of 3-Carbamoyl-5-cyano-4,6-diphenyl-3,4-dihydropyridin- $2(1 \mathrm{H}$ )-one (6a) into 3,5-Dicyano-4,6-diphenyl-3,4-dihydropyridin- $2(1 \mathrm{H}$ )-one ( $\mathbf{1 0 a}$ ).-Compound ( $\mathbf{6 a}$ ) ( 0.6 mmol ) was dissolved in DMF ( 10 ml ) and phosphoryl trichloride ( 0.1 $\mathrm{ml}, 1 \mathrm{mmol}$ ) was slowly added to the continuously stirred solution, using the reaction procedure reported by Bailey et al..$^{22}$ The reaction mixture was then heated at $80^{\circ} \mathrm{C}$ in a waterbath for 10 min . The solution was then allowed to cool and was poured into ice. Aqueous ammonium hydroxide was used to neutralize the aqueous solution, and the reaction product which precipitated was collected by filtration, washed with water, and recrystallized from methanol. Yield $86 \%$.

4,6-Diaryl-3-carbamoyl-5-cyanopyridin-2(1H)-ones (7). General Procedure.-Method (a). To a suspension of the corresponding 3,4-dihydropyridin-2( 1 H )-one (6) in acetic acid ( ca . 10 ml ) at $0^{\circ} \mathrm{C}$ was added a solution of nitrosylsulphuric acid at $0{ }^{\circ} \mathrm{C}$, prepared from sodium nitrite ( 5 mmol ) in sulphuric acid $(50 \mathrm{ml})$ and water $(16 \mathrm{ml}) .^{23.24}$ The reaction mixture was stirred until total solution occurred and the mixture was then kept at room temperature until no starting material remained (t.l.c.) (about $10-12 \mathrm{~h}$ ). The solution was poured into crushed ice and the resulting solid was filtered off and recrystallized from methanol.

Method (b). The corresponding 6-hydroxypiperidin-2-one (5) was subjected to treatment similar to the one described in method (a), but the reaction time was $c a .24 \mathrm{~h}$.

3-Carbamoyl-5-cyano-4,6-diphenylpyridine-2(1H)-one (7a). $82 \%$ yield by method (a) and $95 \%$ by method (b), m.p. 301$302{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 72.1 ; \mathrm{H}, 4.3 ; \mathrm{N}, 13.5 . \mathrm{C}_{19} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{2}$ requires C , $72.38 ; \mathrm{H}, 4.12 ; \mathrm{N}, 13.33 \%$ ); $v_{\text {max. }} 3430,3270,3100-2500$, $2220,1660,1590,1565,1540$, and $1490 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 7.00-7.60$ ( $12 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ and $\mathrm{NH}_{2}$ ).

3-Carbamoyl-5-cyano-6-phenyl-4-(p-tolyl)pyridin-2(1H)-one ( 7 b ). $92 \%$ yield by method (a) and $90 \%$ yield by method (b), m.p. $272-273{ }^{\circ} \mathrm{C}$ (Found: C, 72.7; H, 4.4; N, 12.7. $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2}$ requires $\mathrm{C}, 72.94 ; \mathrm{H}, 4.55 ; \mathrm{N}, 12.76 \%$ ); $v_{\text {max. }} 3620,3430,3200-$ $2500,2220,1735,1640,1610,1560,1540,1490$, and 1440 $\mathrm{cm}^{-1} ; \delta_{\mathrm{H}} 6.90-7.70\left(11 \mathrm{H}, \mathrm{m}, \mathrm{ArH}\right.$, and $\left.\mathrm{NH}_{2}\right)$ and $2.30(3 \mathrm{H}, \mathrm{s})$ $\mathrm{CH}_{3}$ ).

3-Carbamoyl-4-(p-chlorophenyl)-5-cyano-6-phenylpyridin$2(1 \mathrm{H})$-one ( 7 c ). $85 \%$ yield by method (a) and $93 \%$ yield by method (b), m.p. $280-281{ }^{\circ} \mathrm{C}$ (Found: C, 64.9; H, 3.3; N, 11.7 ; Cl , 9.9. $\mathrm{C}_{19} \mathrm{H}_{12} \mathrm{ClN}_{3} \mathrm{O}_{2}$ requires C, $65.23 ; \mathrm{H}, 3.43 ; \mathrm{N}, 12.01 ; \mathrm{Cl}$, $10.15 \%$ ); $v_{\text {max. }} 3440,3200-2500,2220,1715,1630,1590$, $1560,1485,1455$, and $1435 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 7.33-7.96(11 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$, and $\mathrm{NH}_{2}$ ).

3-Carbamoyl-5-cyano-4-(p-methoxyphenyl)-6-phenylpyridin$2(1 \mathrm{H})$-one ( 7 d ). $67 \%$ yield by method (a) and $75 \%$ yield by method (b), m.p. $246-247^{\circ} \mathrm{C}$ (Found: C, 69.3; H, 4.25; N, 11.9. $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{3}$ requires C, 69.56; $\mathrm{H}, 4.34 ; \mathrm{N}, 12.17 \%$ ); $v_{\text {max. }} 3560$, $3410,3200-2500,2220,1715,1635,1605,1560,1540$, $1520,1485,1470,1460$, and $1440 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 6.70-7.70(11 \mathrm{H}$, $\mathrm{m}, \mathrm{ArH}$, and $\mathrm{NH}_{2}$ ) and $3.70\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{O}\right)$.

3-Carbamoyl-5-cyano-4-(p-nitrophenyl)-6-phenylpyridin$2(1 \mathrm{H})$-one ( 7 e ). $67 \%$ yield by method (a) and $80 \%$ yield by method (b), m.p. $320-321{ }^{\circ} \mathrm{C}$ (Found: C, 63.5; H, 3.6; N, 15.6. $\mathrm{C}_{19} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{4}$ requires C, $63.33 ; \mathrm{H}, 3.33 ; \mathrm{N}, 15.55 \%$ ); $v_{\text {max. }} 3440$, 3 300, $3180,3100-2500,2220,1665,1640,1600,1580$, $1560,1540,1510$, and $1490 \mathrm{~cm}^{-1} ; 7.20-8.30(11 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$, and $\mathrm{NH}_{2}$ ).

4,6-Diaryl-5-cyanopyridin-2(1H)-ones (8). General Pro-cedure.-Concentrated hydrochloric acid ( 100 ml ) was added to a solution of the appropriate 4,6-diaryl-3-carbamoyl-5-cyanopyridin-2 $(1 H)$-one ( 7 ) ( 1.4 mmol ) in dioxane ( $c a .50 \mathrm{ml}$ ). The reaction mixture was heated at reflux temperature for a variable reaction time ( $3-20$ days) until t.l.c. showed no
starting material remaining. The solution was then poured into crushed ice and the precipitate was filtered off, washed with water, and recrystallized.

5-Cyano-4,6-diphenylpyridin-2(1H)-one (8a). Reaction time 8 days, yield $67 \%$, m.p. $281-282^{\circ} \mathrm{C}$ (from methanol) (Found: C, 79.2; H, 4.3; $\mathrm{N}, 10.0 . \mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}$ requires C, 79.41; H, 4.41; N , $10.29 \%$ ); $v_{\text {max. }} 3200-2500,2220,1660,1600,1590,1570$, 1530,1490 , and $1450 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 7.20-7.80(10 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$ and $6.30(1 \mathrm{H}, \mathrm{s}) ; m / z$ (relative intensity) $272\left(M^{+}, 85 \%\right.$ ), 271 ( 100 ), 254 (10), 253 (31), 244 (6), 243 (8), 242 (8), 127 (10), and 126 (8).
5-Cyano-6-phenyl-4-(p-tolyl)pyridin- $2(1 \mathrm{H})$-one ( $\mathbf{8 b}$ ). Reaction time 3 days, yield $81 \%$, m.p. $305-306^{\circ} \mathrm{C}$ (from ethanol) (Found: $\mathrm{C}, 79.7 ; \mathrm{H}, 4.9 ; \mathrm{N}, 9.8 . \mathrm{C}_{19} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}$ requires $\mathrm{C}, 79.72 ; \mathrm{H}$, 4.89; N, 9.79\%); $v_{\max .} 3200-2500,2225,1655,1605,1570$, $1535,1515,1495$, and $1460 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 7.00-7.66(9 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, $6.30(1 \mathrm{H}, \mathrm{s})$, and $2.33\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right)$.
4-(p-Chlorophenyl)-5-cyano-6-phenylpyridin-2(1H)-one (8c). Reaction time 20 days, yield $74 \%$, m.p. $276-277^{\circ} \mathrm{C}$ (from methanol); (Found: C, 69.9; H, 3.3; N, 9.0; Cl, 11.3. $\mathrm{C}_{18} \mathrm{H}_{11} \mathrm{ClN}_{2} \mathrm{O}$ requires $\mathrm{C}, 70.47 ; \mathrm{H}, 3.59 ; \mathrm{N}, 9.14 ; \mathrm{Cl}, 11.58 \%$ ); $v_{\text {max }} 3200-2500,2225,1665,1600,1560,1530,1495$, and $1470 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 7.40-7.90(9 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$ and $6.50(1 \mathrm{H}, \mathrm{s})$.
5-Cyano-4-(p-nitrophenyl)-6-phenylpyridin-2(1H)-one (8). Reaction time 20 days, yield $57 \%$, m.p. $320-321^{\circ} \mathrm{C}$ (from methanol) (Found: C, 68.2; H, 3.7; N, 13.0. $\mathrm{C}_{18} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3}$ requires C, 68.14; H, 3.47; N, 13.25\%); $v_{\text {max. }} 2500-3200,2220$, $1665,1610,1600,1590,1570,1520,1495$, and $1470 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}}$ 7.40-8.60 ( $9 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ) and $6.60(1 \mathrm{H}, \mathrm{s})$.

Transformation of 3,5-Dicyano-4,6-Dicyano-4,6-diphenyl-pyridin- $2(1 \mathrm{H})$-one (11a) into 5-Cyano-4,6-diphenylpyridin$2(1 \mathrm{H})$-one (8a).-Compound (11a) $(0.15 \mathrm{mmol})$ was dissolved in a mixture of ethylene glycol ( 5 ml ) and concentrated hydrochloric acid ( 10 ml ) and the reaction mixture was refluxed for 60 days. T.l.c. showed clearly the transformation of (11a) into the 3 -carbamoyl derivative (7a) and then into the title compound (8a). The solution was poured into crushed ice, and the product was filtered off and recrystallized. Yield $61 \%$.

## 4,6-Diaryl-3-cyano-5-ethoxycarbonyl-6-hydroxypiperidin-

 2-ones (12). General Procedure.-To a suspension of the appropriate ethyl $x$-benzoylcinnamate (4) ( 14 mmol ) in dry methanol (ca. 40 ml ) was added a solution of 2-cyanoacetamide (2) $(14 \mathrm{mmol})$ in the minimal amount of methanol, together with a few drops of piperidine. The reaction mixture was stirred at room temperature for $2-3$ days and the solid which precipitated was then collected by filtration and washed with a small amount of methanol. The resulting compounds were purified by dissolution in methanol at room temperature, followed by cooling the solution at $-25^{\circ} \mathrm{C}$, because compounds (12) decompose at reflux temperature.3-Cyano-5-ethoxycarbonyl-6-hydroxy-4,6-diphenylpiperi-din-2-one (12a). $50 \%$ yield, m.p. $106-108{ }^{\circ} \mathrm{C}$ (Found: C, 69.45; $\mathrm{H}, 5.6 ; \mathrm{N}, 7.3 . \mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{4}$ requires $\mathrm{C}, 69.23 ; \mathrm{H}, 5.49$; N , $7.69 \%$ ); $3650,3400,3300,2250,1725,1670,1595,1475$, and $1450 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 8.76(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 7.00-7.80(10 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.60$ $(1 \mathrm{H}, \mathrm{s}, \mathrm{OH}), 3.00-4.60\left(5 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right.$ and ring Hs$)$, and 0.50 ( $3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}$ ).

3-Cyano-5-ethoxycarbonyl-6-hydroxy-6-phenyl-4-(p-tolyl)-piperidin-2-one (12b). $39 \%$ yield, m.p. $108-109^{\circ} \mathrm{C}$ (Found: C, 69.3; $\mathrm{H}, 5.55 ; \mathrm{N}, 7.5 . \mathrm{C}_{22} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{4}$ requires C, $69.84 ; \mathrm{H}, 5.82 ; \mathrm{N}$, $7.41 \%$ ); $v_{\text {max }} 3655,3400,3290,3100,2250,1725,1675,1590$, and $1515 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 8.60(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 6.70-7.50(9 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, $6.50(1 \mathrm{H}, \mathrm{s}, \mathrm{OH}), 3.10-4.50\left(5 \mathrm{H}, \mathrm{m} \mathrm{CH}_{2}\right.$ and ring Hs$), 2.16$ (3 $\left.\mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right)$, and $0.50\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right)$.

4-(p-Chlorophenyl-3-cyano-5-ethoxycarbonyl-6-hydroxy-6-phenylpiperidin-2-one ( 12 c ). $31 \%$ yield, m.p. $107-108{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 62.9 ; \mathrm{H}, 4.9 ; \mathrm{N}, 6.85 ; \mathrm{Cl}, 9.2 . \mathrm{C}_{21} \mathrm{H}_{19} \mathrm{ClN}_{2} \mathrm{O}_{4}$ requires

C, 63.23; H, 4.76; N, 7.02; Cl, $8.91 \%$ ); $v_{\text {max. }} 3660,3400,3310$, 3 180, $2250,1720,1675,1595,1495$, and $1475 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 8.80(1$ $\mathrm{H}, \mathrm{s}, \mathrm{NH}), 7.00-7.60(9 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.70(1 \mathrm{H}, \mathrm{s}, \mathrm{OH}), 3.20-$ $4.60\left(5 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right.$ and ring Hs$)$, and $0.53\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right)$.

## 4,6-Diaryl-3-cyano-5-ethoxycarbonyl-3,4-dihydropyridin-

 2(1H)-ones (13). General Procedure.-A suspension of the appropriate compound (12) ( 2 mmol ) in $75 \%$ sulphuric acid ( ca . 10 ml ) was stirred at room temperature for 30 min . The reaction mixture was then poured into ice-water and the resulting precipitate was washed with water and recrystallized from aqueous ethanol.3-Cyano-5-ethoxycarbonyl-4,6-diphenyl-3,4-dihydropyridin$2(1 \mathrm{H})$-one (13a). $89 \%$ yield, m.p. $150-151^{\circ} \mathrm{C}$ (lit., ${ }^{5} 151$ $153{ }^{\circ} \mathrm{C}$ ).

3-Cyano-5-ethoxycarbonyl-6-phenyl-4-(p-tolyl)-3,4-dihydro-pyridin- $2\left(1 \mathrm{H}\right.$ )-one ( 13 b ). $82 \%$ yield, m.p. $180-181^{\circ} \mathrm{C}$ (lit., ${ }^{5}$ $179-181^{\circ} \mathrm{C}$ ).

4-(p-Chlorophenyl)-3-cyano-5-ethoxycarbonyl-6-phenyl-3,4-dihydropyridin- $2\left(1 \mathrm{H}\right.$ )-one ( 13 c ). $65 \%$ yield, m.p. $163-164{ }^{\circ} \mathrm{C}$ (lit., ${ }^{5}$ 164-165 ${ }^{\circ} \mathrm{C}$ ).

4,6-Diaryl-3-cyano-5-ethoxycarbonylpyridin-2(1H)-ones (14). General Procedure.-Method (a). The appropriate 4,6-diaryl-3-cyano-5-ethoxycarbonyl-6-hydroxypyridin-2(1H)-one (13) (2 mmol ) was dissolved in the minimal amount of pyridine. The solution was cooled to $-10^{\circ} \mathrm{C}$ and thionyl chloride ( 20 ml ) was added dropwise to the magnetically stirred solution. The mixture was kept for a further 30 min and was then poured into crushed ice. The resulting precipitate was collected by filtration and purified by recrystallization in ethanol.

Method (b). To a magnetically stirred suspension ( 0.14 mmol ) of the corresponding 4,6-diaryl-3-cyano-5-ethoxycarbonyl-3,4-dihydropyridin- $2(1 H)$-one (13) in acetic acid ( 0.4 ml ) was added a solution of nitrosylsulphuric acid [prepared from sodium nitrite ( 0.2 mmol ), sulphuric acid ( 2 ml ), and water ( 0.6 ml )] dropwise. After 15 min the reaction mixture was poured into icewater. The precipitate was filtered off, washed with water, and recrystallized.

3-Cyano-5-ethoxycarbonyl-4,6-diphenylpyridin-2(1H)-one (14a). $78 \%$ yield by method (a) and $80 \%$ yield by method (b), m.p. $261-262^{\circ} \mathrm{C}$ (lit., ${ }^{5} 256-257^{\circ} \mathrm{C}$ ).

3-Cyano-5-ethoxycarbonyl-6-phenyl-4-(p-tolyl)pyridin$2(1 \mathrm{H})$-one ( 14 b ). $94 \%$ yield by method (a), m.p. $295-296^{\circ} \mathrm{C}$ (lit., ${ }^{5} 300^{\circ} \mathrm{C}$ ).

4-(p-Chlorophenyl)-3-cyano-5-ethoxycarbonyl-6-phenyl-pyridin- $2(1 \mathrm{H}$ )-one ( 14 c ). $78 \%$ yield by method (a), m.p. $282-$ $284{ }^{\circ} \mathrm{C}$ (Found: C, 66.4; H, 4.1; N, 7.8; Cl, 9.3. $\mathrm{C}_{21} \mathrm{H}_{15} \mathrm{ClN}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 66.52 ; \mathrm{H}, 3.94 ; \mathrm{N}, 7.47 ; \mathrm{Cl}, 9.45 \%$ ); $v_{\text {max }} 3450,3400-$ $2700,2220,1700,1640,1600,1590$, and $1490 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 7.20-$ $8.10(9 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 3.60\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{2}\right)$, and $0.63\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right)$.

3-Cyano-5-ethoxycarbonyl-4-(p-nitrophenyl)-6-phenyl-pyridin- $2(1 \mathrm{H})$-one ( 14 e ). $94 \%$ yield by method (a), m.p. $270-$ $271{ }^{\circ} \mathrm{C}$ (lit.,,$^{5} 270-271^{\circ} \mathrm{C}$ ).

5-Cyano-6-oxo-2,4-diphenyl-3,4,5,6-tetrahydropyridine-3carboxylic Acid (15a).-3-Cyano-5-ethoxycarbonyl-6-hydroxy-4,6-diphenylpiperidin-2-one (12a) ( 3 mmol ) was suspended in $98 \%$ sulphuric acid $(10 \mathrm{ml})$ and the reaction mixture was stirred at room temperature for 5 h . The resulting solution was poured into ice-water and the solid which precipitated was filtered off and washed with cold water. The mother liquors were extracted with chloroform, and the extract was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated. The residue was added to the first crop of solid, and the whole was recrystallized from ethanol. Yield $82 \%$, m.p. 262- $264^{\circ} \mathrm{C}$.

The title compound was also obtained by similar treatment of 3-cyano-5-ethoxycarbonyl-4,6-diphenyl-3,4-dihydropyridin-
$2(1 \mathrm{H})$-one (13a), yield 70\% (Found: C, 71.6; H, 4.65; N, 8.8. $\mathrm{C}_{19}$ $\mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires C, 71.50; H, 4.40; N, 8.80\%); $v_{\text {max. }} 3300-$ $2550,1700,1675,1625,1595,1510$, and $1450 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 11.2(1$ $\mathrm{H}, \mathrm{s}, \mathrm{NH}), 7.1-7.6(10 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 4.2(1 \mathrm{H}, \mathrm{d})$, and $3.5(1 \mathrm{H}, \mathrm{d})$; $m / z$ (rel. int. 318 ( $M^{+}, 4 \%$ ), 276 (6), 275 (35), 274 (100), 273 (7), 272 (6), 257 (5), and 256 (19).

Attempted Reaction of Ethyl $\alpha$-Benzoylcinnamate (4a) with Malonamide (1).-To a suspension of ethyl $\alpha$-benzoylcinnamate (4a) ( 3.6 mmol ) in dry ethanol ( 10 ml ) was added a suspension of malonamide (1) ( 3.6 mmol ) in dry methanol ( 5 ml ), together with a few drops of piperidine. The reaction mixture was stirred at room temperature for a long time without any progress. The starting materials were also recovered when the reaction was performed at room temperature or with sodium ethoxide as the basic catalyst. If a molar amount of sodium ethoxide was used, or the reaction was performed without any solvent at $200^{\circ} \mathrm{C}$, only an intractable complex mixture was obtained.

## Acknowledgements

Support of this work by a Grant of the Comisión Asesora de Investigación Científica y Técnica of Spain is gratefully acknowledged. Two of us (P. Z. and M. J. R.) thank the Ministerio de Educación y Ciancia of Spain for a fellowship.

## References

1 Part 45, L. Fuentes, J. Vaquero, J. Castillo, M. Ardid, and J. L. Soto, Heterocycles, 1985, 23, 93.
2 C. Seoane, J. L. Soto, and M. P. Zamorano, Heterocycles, 1980, 14, 639.

3 'Comprehensive Organic Chemistry' ed. D. H. R. Barton, Pergamon, Oxford, 1979, vol. 4.

4 G. R. Newkome and W. P. Paudler. 'Contemporary Heterocyclic Chemistry,' Wiley-Interscience, New York, 1982.
5 C. Seoane, J. L. Soto, P. Zamorano, and M. Quinteiro, J. Heterocycl. Chem., 1981, 18, 309.
6 H. Junek and A. Schmidt, Monatsh. Chem., 1967, 98, 1097.
7 Z. Bomika, M. Andaburskaya, and J. Duburs, Khim. Geterotskl. Soedin., 1975, 1108 (Chem. Abstr., 1975, 83, 193035k).
8 M. Haffer, U.S.P. 2440 045, 1946 (Chem. Abstr., 1946, 40, 5073 ${ }^{8}$ ).
9 J. A. Ciller, C. Seoane, and J. L. Soto, An. Quim., 1980, 76(C), 281.
10 L. J. Bellamy, 'The Infrared Spectra of Complex Molecules,' Chapman and Hall, London, 1975, 3rd edn.
11 F. Ramirez and A. P. Paul, J. Am. Chem. Soc., 1955, 77, 1035; J. A. Gibson, W. Kynaston, and A. S. Lindsey, J. Chem. Soc., 1955, 4340.
12 J. L. Soto, C. Seoane, J. Valdés, and N. Martin, An. Quim., 1979, 75, 152.

13 C. A. Haasnoot, F. A. M. de Leeuw, and C. Altona, Tetrahedron, 1980, 36, 2783.
14 M. L. Huggins, J. Am. Chem. Soc., 1953, 75, 4123.
15 H. Kaufman, Ber., Dtsch. Chem. Ges., 1917, 50, 5150.
16 D. Drapkina, V. Brudz, and B. Bolotin, Tr. Vses. Nauchno-Issled. Inst. Khim. Reakt. Osobo Chist. Khim. Veshchestv., 1967, 30, 333 (Chem. Abstr., 1968, 69, 6727p).
17 S. Ruheman, J. Chem. Soc., 1903, 83, 717.
18 J. P. Vecchionacci, J. C. Ganevet, and Y. Graff, Bull. Soc. Chim. Fr., 1974, 1683.
19 I. Gosney and D. Lloyd, Tetrahedron, 1973, 29, 1697.
20 H. L. Holmes and D. J. Currie, Can. J. Chem., 1969, 47, 4076.
21 Ch. Eby and Ch. Hauser, J. Am. Chem. Soc., 1957, 79, 723.
22 A. S. Bailey, B. R. Henn, and J. Logdom, Tetrahedron, 1963, 19, 161.
23 H. H. Hodgson and A. P. Mahadevan, J. Chem. Soc., 1947, 325.
24 L. Fieser and M. Fieser, 'Reagents for Organic Synthesis,' Wiley, New York, 1967, p. 755.

